SYNTHESIS OF ISOMERIC PAIRS OF IMIDAZOLE N-OXIDES

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Several & -halogenketones have been found to react with hydroxylamine to yield & -hydroxylaminooximes along with & -dioximes (1). The configuration of the formers has been established by comparing their ultraviolet absorption spectra with those of related oximes of the known configuration. The configuration of anti-isomers was confirmed by their ability to form Fe and Cu complexes.

Condensation of anti-alkyl forms of N-(1-hydroxylmino-1-phenylpropyl-2)hydroxylamine (I,m.p.164.0-166.0°) and N-(1-hydroxylmino-2,3-benzocycloheptyl-7)hydroxylamine (II,m.p. 187.0-188.0°) with acetaldehyde resulted in 1-hydroxy-2,5-dimethyl-4-phenyl-3-imidazoline-3-oxide (III,m.p.93.5-94.5°, 80%) and in 1-hydroxy-2-methyl-[2',3'-benzocycloheptyl-(7',1':5,4)-3-imidazoline-3-oxide] (IV,m.p.141.0-142.0°), respectively. III and IV refluxed with acetic or benzoic anhydrides in chloroform gave acyl derivatives of III and IV (V,m.p.85.5-86.0°; VI.m.p.100.0-100.5°; VII,m.p.81.5-82.0°).

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Heating of V or VI in vacuo afforded a product (VIII, m.p.149.0-150.0°) empirical formula of which (C11H12N2O) was in accordance with the elimination of a molecule of carboxylic acid from V or VI. The ultraviolet spectra of VIII(λ max 252 mm (logs 4.01)) shows no broad band at 290-296 mm (logs 4.2-4.3) characteristic of C6H5-C=N+O chromophore in spectra III. IV and V. The n.m.r. spectrum of VIII taken on JNM-3 spectrometer (40 Mc) using a 30% CDCl3 solution consists of the three sharp singlets (cf.(2)), an area ratio being about 6:5:1. The peak at 8.00 (corresponds to six hydrogen atoms of two methyl groups, the peak at 2.75 to five hydrogen atoms of benzene ring and the peak at -5.171 indicates that we were dealing with a strong hydrogen bond. The treatment of VIII with phosphorus trichloride in chloroform yielded 2,4(5)-dimethyl-5(4)-phenylmidazole (IX) identified by infrared spectrum and by a melting point of a mixture with authentic sample (3). On the basis of these facts we concluded that VIII had the structure of 2.5-dimethyl-4-phenylimidazole-3-oxide or its tautomeric form carrying hydroxygen function in 3-position.

The determination of the molecular weight of VIII by the isothermic distillation method in chloroform solution indicated that VIII has been appreciably associated. At a concent-

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ration of 10% it exhibited an apparent melecular weight of 7-9 times of the formula weight. These findings were confirmed by noting one of the n.m.r. signals in a very low field.

When ethanol solution of V or VI was treated with dry hydrogen chloride hydrochloride VIII (m.p.205.0-206.0°) was formed. However, the action of hydrogen chloride upon ethanol solution of III gave also rise to hydrochloride I (m.p.213.0-215.00), mixed melting point of which with hydrochloride VIII showed depression. The elemental analysis of free base (X,m.p.162.0-163.00) indicated the composition of VIII (C44H40N20) while these compounds differed in infrared and ultraviolet spectra. The ultraviolet spectrum of X ($\lambda \frac{\text{EtOH}}{\text{max}}$ 270 mm (loge 4.07)) does not posses a broad band at 290-296 mm indicating the absence of C_6H_5 -C=N+O group. The n.m.r. spectra of X is consistent with that of VIII both in position of bands (7.88, 2.85 and -5.781) and the area ratio. I reacted with phosphorus trichloride to yield IX. The above evidence suggested that X and VIII differed only in the position of the oxygen which might be attached to one of the two of nitrogen atoms.

The n.m.r. spectrum and the molecular weight determination of XI point to the same degree of association as VIII.

VII and IV when treated under similar conditions resul-

ted in 2-methyl- [2,3-benzocycloheptyl-(7,1:5,4)-imidazole-3-axide] (XI.m.p.232.6-233.0°) and 2-methyl- [2,3-benzocycloheptyl-(7,1:4,5)-imidazole-3-oxide] (XII.m.p.212.0-213.0°), respectively.

XI and XII reacted with phosphorus trichloride to form 2-methyl-[2,3-benzocycloheptyl-(7,1:4,5)-imidazole] (XIII hydrochloride,m.p.265.0-267.0°).

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